## Unit Conversions

## Length

$$
\begin{array}{lll}
1 \mathrm{ft}=12 \mathrm{in} \text { (exact) } & 1 \mathrm{yd}=3 \mathrm{ft} \text { (exact) } & 1 \mathrm{mi}=5280 \mathrm{ft} \text { (exact) } \\
1 \mathrm{mi}=1.609 \mathrm{~km} & 1 \mathrm{in}=2.54 \mathrm{~cm} \text { (exact) } &
\end{array}
$$

## Mass*

$1 \mathrm{lb}=16 \mathrm{oz}$ (exact)
$1 \mathrm{lb}=0.4536 \mathrm{~kg}$
$1 \mathrm{u}=1.66054 \times 10^{-27} \mathrm{~kg}$

## Volume

$$
\begin{aligned}
& 1 \mathrm{gal}=4 \mathrm{qt}(\text { exact }) \\
& 1 \mathrm{~L}=1.057 \mathrm{qt}
\end{aligned} \quad 1 \mathrm{~mL}=1 \mathrm{~cm}^{3} \text { (exact) } \quad 1 \mathrm{gal}=3.785 \mathrm{~L}
$$

## Energy

$$
1 \mathrm{cal}=4.184 \mathrm{~J} \text { (exact) } \quad 1 \mathrm{Cal}=1000 \mathrm{cal} \text { (exact) }
$$

## Pressure

$$
\begin{array}{ll}
1 \mathrm{~atm}=760 \text { Torr (exact) } & 1 \text { Torr }=1 \mathrm{mmHg} \text { (exact) }{ }^{* *} \\
1 \mathrm{~atm}=101,325 \mathrm{~Pa} \text { (exact) } & 1 \text { bar }=1 \times 10^{5} \mathrm{~Pa} \text { (exact) }
\end{array}
$$

## Time

$$
1 \text { day }=24 \mathrm{hr}(\text { exact }) \quad 1 \mathrm{~min}=60 \mathrm{sec}(\text { exact })
$$

* In the English system, the pound is defined with respect to the kilogram at the standard value for acceleration, $9.80665 \mathrm{~m} / \mathrm{s}^{2}$.
**The relationship is not exact but for our purposes the difference is negligible.
Lide, David R., Ed., Handbook of Chemistry and Physics, $83^{\text {rd }}$ ed.; CRC Press: Boca Raton FL, 2002, 134 to 1-45.

